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# **Photodetachment of $\text{SF}_6^-$ and of small $(\text{SF}_6)_n^-$ cluster anions**

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## Abstract

In this work  $\text{SF}_6^-$  and small  $(\text{SF}_6)_n^-$  cluster anions are mass selected and trapped in an electrostatic ion beam trap to be interrogated by photodetachment with tunable wavelength ns laser pulses. Photodetachment efficiency curves are constructed by detecting the number of neutral photodetachment products as well as the number of trapped ions as a function of photon energy. With the advent of fast ion beam methods for detection of neutral photo-products and efficient overlap in a co-linear geometry of the laser beam and electrostatically trapped ion beam, we extend the range of  $\text{SF}_6^-$  photodetachment efficiency measurements towards low photon energies in the range of 420-700nm that are below the detection limit of the previously used method. The shape of the photodetachment efficiency curve is found to be monotonically increasing; in agreement with the curve shape expected based on photoelectron spectroscopy data, indicating no significant contributions of indirect photodetachment processes in the recorded photon energy range. Directly comparing the measured photodetachment efficiency curves for  $\text{SF}_6^-$  and for small clusters formed by addition of one or two  $\text{SF}_6$  molecules, we deduce that the negative charge is localized on one of the  $\text{SF}_6$  units within the cluster. We also describe a compact photofragment spectrometer that was designed, constructed and installed inside the electrostatic ion beam trap as a part of this M.Sc. work. In future experiments this new device will allow to pursue also differential photodissociation measurements in addition to the already described photodetachment efficiency curve measurements.



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## 1. Background

The work presented here explores the photodetachment of  $\text{SF}_6^-$  molecular anion and of small  $(\text{SF}_6)_n^-$  cluster anions. In addition to the basic research interests and applications in accelerator technology, where sulfur hexafluoride is utilized for its high dielectric constant,<sup>1</sup>  $\text{SF}_6$  is used in many industrial applications. Its relative inertness and large cross section for electron attachment<sup>2</sup> make  $\text{SF}_6$  an efficient electron scavenger used in electrical industry<sup>3</sup>, plasma etching processing<sup>4,5</sup> and manufacturing of thin film transistor liquid crystal display.<sup>6</sup> Although  $\text{SF}_6$  is an inert gas it is one of the most potent greenhouse gases released to the earth's atmosphere due to its strong IR absorption<sup>7</sup> and its long lifetime in Earth's atmosphere.<sup>8</sup> Interestingly the dominant  $\text{SF}_6$  destruction mechanism in the atmosphere is believed to be dissociative electron attachment. Hence its atmospheric lifetime estimates depend strongly on the assumed cross-sections and electron density in the stratosphere, with estimates varying between 10000 years when neglecting dissociative electron attachment and 400 years when assuming each collision event destroys the molecule.<sup>8</sup> As the 100 years global warming potential of  $\text{SF}_6$  is 23,900 higher than that of  $\text{CO}_2$ ,<sup>9</sup>  $\text{SF}_6$  emissions to the atmosphere are monitored and were shown to grow from  $\sim 0.75\text{Gg/yr}$  to  $\sim 7.5\text{Gg/yr}$  between 1973 to 2008,<sup>10</sup> compatible with the reported growth of  $\text{SF}_6$  sales.<sup>11</sup> Although extensive research efforts were made in the past decade, there is still debate about some of the fundamental properties of the  $\text{SF}_6$  molecule.<sup>12</sup> In particular, experimental measurements of the adiabatic electron affinity ( $\text{EA}_{\text{ad}}$ ) report values between 0.55-1.20eV.<sup>13-16</sup> While theoretical calculations also result in a wide range of electron affinities 0.73-1.03eV.<sup>12,17</sup> The reported lifetimes of hot metastable  $\text{SF}_6^-$  anions that are formed by electron attachment vary by many orders of magnitude from  $2\mu\text{s}$  to 1.1sec depending on experimental method.<sup>18-21</sup> Recent storage ring measurements with long storage times showed a non exponential decay of  $\text{SF}_6^-$  anions with  $t^{-1.5}$ , extending up to 2sec indicating that the instantaneous decay rate slows down as a function of time.<sup>21</sup> Figure 1. Shows a schematic sketch of the potential curves for the neutral  $\text{SF}_6$  and its  $\text{SF}_6^-$  anion. The addition of an electron significantly changes the equilibrium S-F bond length from  $1.56\text{\AA}$  to  $1.71\text{\AA}$ ,<sup>22</sup> resulting in a very different vertical electron detachment energy of about 3.1eV,<sup>23</sup> and adiabatic electron affinity( $\text{EA}_{\text{ad}}$ ) of about 1 eV.<sup>12</sup> The large difference leads to a vanishing transition probability from the anion ground state directly to the neutral ground state, consequently resulting in experimental difficulties in  $\text{EA}_{\text{ad}}$  determination. Photoelectron spectroscopy is usually a good method for determining electron binding energies, recent measurements of cold  $\text{SF}_6^-$  showed unexpected double progression in the photoelectron spectrum<sup>24</sup> that inspired accurate structure calculations. The new calculations indicate that  $\text{SF}_6^-$  departs from the previously assumed  $\text{O}_h$  symmetry<sup>22</sup> of  $\text{SF}_6$  and has a  $\text{C}_{4v}$  symmetry, providing an explanation for the observed double progression.<sup>17</sup> The inset of figure 1 shows the suggested  $\text{SF}_6^-$  structure with  $\text{C}_{4v}$  symmetry, when the elongated S-F bond length is about  $1.9\text{\AA}$  and the short S-F bond equilibrium is  $1.6\text{\AA}$  and the other four S-F bonds are about  $1.70\text{\AA}$  long. The dissociation energy of  $\text{SF}_6$  towards  $\text{SF}_5+\text{F}$  is about  $\sim 3.9\text{eV}$ .<sup>25</sup> On the other hand,  $\text{SF}_6^-$  dissociation energies towards  $\text{SF}_5^-+\text{F}$  and  $\text{SF}_5+\text{F}^-$  are  $\sim 1.44\text{eV}$ <sup>12</sup> and  $1.75\text{eV}$  respectively,<sup>26</sup> which also lie in the range between the adiabatic detachment and vertical detachment energies. Consequently one can expect  $\text{SF}_6^-$  to exhibit interesting

competition between detachment and dissociation following absorption of a photon in this energy range. The same competition between detachment and dissociation plays a critical role also in deciding the fate of the  $\text{SF}_6^-$  formed in electron attachment to  $\text{SF}_6$  in the atmosphere. Detachment will return the system back to the neutral molecule, while dissociation will remove  $\text{SF}_6$  from the atmosphere.

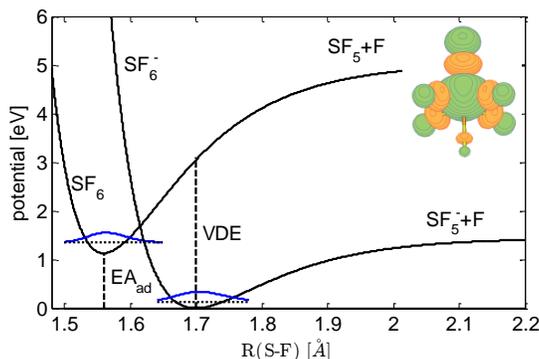


Figure 1: schematic sketch of  $\text{SF}_6$  and  $\text{SF}_6^-$  molecules potential curve (solid black) and their ground state wave functions (solid blue), illustrated by vertical dashed lines the adiabatic and vertical detachment electron energies ( $EA_{ad}$ , VDE respectively). in the inset a suggested  $\text{SF}_6^-$  structure with  $C_{4v}$  symmetry.<sup>17</sup>

To help consider the  $\text{SF}_6^-$  photodetachment, we first explain the photoelectron spectroscopy (PES) measurement performed by Bopp *et al.*<sup>24</sup> and then explain our complementary PDE measurement. In a PES experiment, a photon in the UV is absorbed by the ground state molecular anion and the energy spectrum of emitted photoelectron is recorded. The measured electron kinetic energy can be related, by the energy conservation law, to the energy difference between the anionic ground state and excited neutral molecules:

$$(1) \quad \Delta E = h\nu - E_k$$

Where  $\Delta E$  is the difference between the neutral excited state energy and the anion ground state,  $h\nu$  is the photon energy and  $E_k$  is the electron kinetic energy.

Figure 2 shows a simplified one dimensional harmonic oscillator model that helps considering the PES process. When equilibrium bond lengths are taken to be 1.56Å and 1.70Å for the neutral and anion respectively, the  $EA_{ad}$  is 1eV and the vertical electron detachment energy is 3.1eV. Considering instantaneous photo-detachment from the anionic ground state to the free electron continuum, we can assume the nuclei did not have time to move. Accordingly, the relative probability of producing the different neutral vibrational states can be estimated by the Franck-Condon (FC) approximation that the transition probability will be proportional to the square of the overlap integral of the initial and final vibrational wave function, also known as the FC factor:

$$(2) \quad FC(v) = \left| \int_{-\infty}^{\infty} \psi_v(x-x_{neut})^* \psi_0(x-x_{anion}) dx \right|^2$$

Where  $x_{neut}$ ,  $x_{anion}$  are the equilibrium bond lengths, and  $\psi_v$  is the vibrational wave function of the  $v$  state as shown in eq. 3:

$$(3) \quad \psi_v(x) = \frac{1}{\sqrt{2^v v!}} \left( \frac{m\omega}{\pi\hbar} \right)^{\frac{1}{4}} e^{-\frac{m\omega x^2}{2\hbar}} H_v \left( \sqrt{\frac{m\omega}{\hbar}} x \right)$$

Where  $v$  is the vibration number,  $m$  is the reduced mass of the nuclei,  $\omega$  is the angular frequency of the harmonic oscillator and  $H_v$  are Hermit polynomials.

The calculated FC factor for the harmonic oscillator model is shown in figure 2 as a function of photon energy. As mentioned before, since the equilibrium bond length is shift by 0.14Å the overlap between the two ground states vanishes. On the other hand, the overlap reaches its maximal value when the energy difference between the final and initial state equals to the potential curve difference at the  $SF_6^-$  equilibrium distance. The characteristic vibrational wavefunction illustrated in figure 2 shows the good overlap of the outer lobe of the excited vibrational state and the anion ground state.

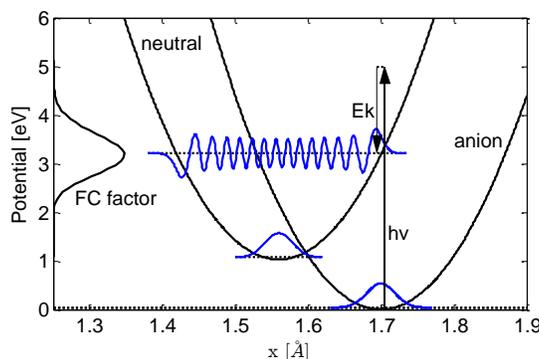


Figure 2: An illustration of photo electron spectroscopy. Scheme of two harmonic oscillator potential, neutral and anion molecule. a photon in the UV (solid arrow) hit the anion, change it to a certain vibration of the neutral molecule by the FC factor (on the left), and the dashed arrow size represent the emitted electron kinetic energy, the end of the dashed arrow is final vibration high and the x-location represent the place in the vibration at the transition.

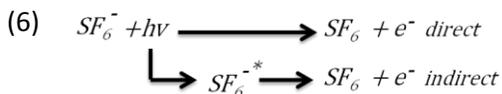
In the experiments described here we perform a complementary measurement to PES, in which we measure the photo detachment efficiency curve as a function of photon energy (PDE). Figure 3 describes schematically the scenario: a photon is absorbed by the anion, resulting in a superposition of final states including an emission of an electron with different finite kinetic energies and the corresponding energetically allowed final vibrational states of the neutral molecule. As previously described, we can estimate the probability for direct transition to each final state according to the FC principle. Therefore, as illustrated in figure 3, the change in the total detachment probability should be proportional to the sum over all FC factors of the energetically allowed final states, i.e. the states that lie below the energy of the absorbed photon.

$$(4) \quad PDE \propto \sum_{v=0}^{v_{max}} FC(v)$$

Keeping in mind that the PES spectrum is proportional to the FC factor, one can write the PDE as a function of energy.

$$(5) \quad PDE(h\nu) = \int_0^{h\nu} PES(E) dE$$

When  $h\nu$  is the photon energy. However, contrary to PES, PDE can be sensitive also to a possible contribution of indirect detachment mechanisms. i.e. photoabsorption into an excited anion state followed by autoemission as illustrated in equation 6.



Analysis of electron attachment experiments,<sup>27,28</sup> and theoretical calculations,<sup>29</sup> indicate the existence of such excited  $SF_6^-$  states that may in principle contribute to the PDE in the relevant energy range between the vertical and adiabatic detachment energies. As opposed to the monotonously increasing PDE of a direct photodetachment mechanism illustrated in figure 3, indirect contribution can be expected to produce absorption resonances, corresponding to the discrete states of the excited anion.

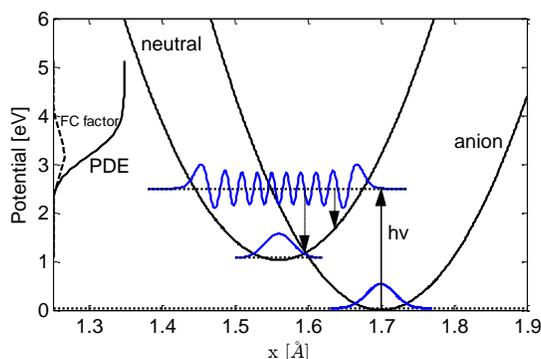


Figure 3: An illustration of photodetachment spectroscopy. approximation of neutral and anion potentials to harmonic oscillator (solid black), the ground states wave function (solid blue), a photon hit the anion molecule (solid blue arrow), detached an electron and raise to a superposition of all the vibrational levels until the max vibration allows by the photon energy (dashed arrows), and the neutral fragment is measured, on the left side the normalized FC factor (dashed black) and the PDE (solid black) as a function of photon energy.

Figure 4 shows the previously measured PDE of  $SF_6^-$  on an absolute cross section scale as function of photon wavelength, reproduced from the work of Mock *et al.*<sup>30</sup> The cross section abruptly vanishes below the experimental detection limit for wavelengths above 400nm corresponding to photon energies below 3.1eV. In my M.Sc. project I extended the  $SF_6^-$  PDE measurement beyond the previously available range of 300-400nm to the 420-705 nm wavelengths, corresponding to the 2.94-1.76eV range of photon energies.

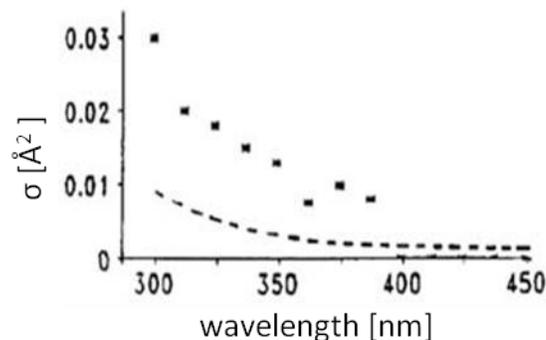


Figure 4:  $\text{SF}_6^-$  absolute cross section by Mock<sup>30</sup>

Furthermore, harmonic oscillator model calculations shown in figure 5 demonstrate how the PDE in this range is sensitive to even tiny  $0.01\text{\AA}$  changes in the anion equilibrium bond length, changing both overall magnitude and shape of the PDE curve. I therefore explored also the PDE of small cluster anions based on the  $\text{SF}_6^-$  system of the  $(\text{SF}_6)_n^-$  form.<sup>31,32,26</sup> Aiming to determine the effect of the interaction of the  $\text{SF}_6^-$  anion with the nearby neutral  $\text{SF}_6$  molecules on its effective bond length. Although neutral  $(\text{SF}_6)_n$  clusters are bound by extremely weak Van der Waals, the additional charge changes the induced dipole – induced dipole interactions into a charge – induced dipole interaction. For similarly interacting positively charged rare gas clusters, measurements established that the charge is delocalized over the rare gas cluster, resulting for example in clusters of the form of  $\text{Xe}_n(\text{Xe}_3)^+$  in the case of Xe atoms.<sup>33,34</sup> If such delocalization occurs in  $(\text{SF}_6)_n^-$  clusters it is bound to modify the effective S-F bond length and strongly affect our PDE curve measurements.

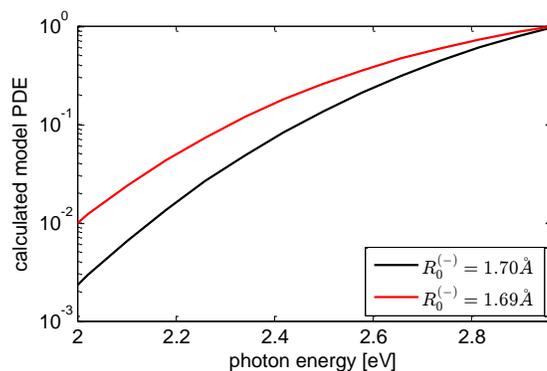


Figure 5: An illustration that come to show the different in the shape for normalized numerical calculation PDE, for a change of  $0.01\text{\AA}$  in equilibrium bond length, from  $1.69\text{\AA}$  (red) to  $1.70\text{\AA}$  (black).

In addition to the PDE measurement setup, I designed and constructed a compact photofragment spectrometer that will enable us to detect and differentiate photo-dissociation products according to their charge over mass ratio and map the competition between detachment and dissociation in this as well as in other molecular anion systems.

## 2. Experimental setup

Our experimental setup, shown schematically in figure 6, is designed to prepare cold molecular and cluster anions, mass select them and record their photodetachment efficiency as a function of photon energy. Briefly speaking, cold molecular and cluster anions are formed in a pulsed supersonic expansion "Even-Lavie" ion source.<sup>35</sup> The formed ions are then accelerated by 4.2keV and directed towards the Electrostatic Ion Beam Trap (EIBT),<sup>36-39</sup> a resonator composed of two electrostatic mirrors that can trap fast ions oscillating back and forth between the two mirrors. Every experimental cycle, a bunch of ions is injected into the trap by lowering the potential ( $V_p$ ) on one of the electrodes forming the entrance mirror, thus allowing ions to enter the trap. After the desired ions enter the trap, the potential is raised and effectively traps the ions by closing the "hole" in the mirror. Under UHV conditions (typically  $10^{-10}$  Torr) the ions oscillate in the EIBT for times as long as 3s and more. As the ion oscillation frequency is proportional to the  $\sqrt{q/mass}$ , applying a synchronized periodic  $\pm 100$  voltage pulse to a deflector (labeled  $V_{KO}$ ) located inside the trap allows selecting a specific ion species by "kicking out" ions that oscillate at different frequencies.<sup>40</sup> The trapped ions are monitored by observing neutral products of collisions with the residual gas that exit the trap and impinge on an MCP detector located behind the exit mirror. Furthermore, a pickup ring electrode located near the center of the trap allows following the ions directly by observing the image charge induced on the pick-up electrode by the periodic passage of the charged particles.<sup>37,41</sup> A 3mm hole in the center of the MCP detector allows us to efficiently overlap the trapped ion beam with a co-linear laser beam of a pulsed tunable wavelength OPO laser<sup>41</sup>. Neutral products of the laser-anion interaction exit the ion trap and are detected on the MCP outside the trap at a time delay after the laser pulse, corresponding to the velocity of the trapped ions and the product mass. Thus by recording the yield of neutral products and laser power as function of wavelength we are able to construct the photodetachment efficiency curve. Furthermore, as will be explained in section 3.3, applying a high voltage potential on the deflector electrodes in the central region of the trap will allow us to detect also charged photodissociation products and determine their branching fractions by resolving the photo-products by their charge over mass ratio.

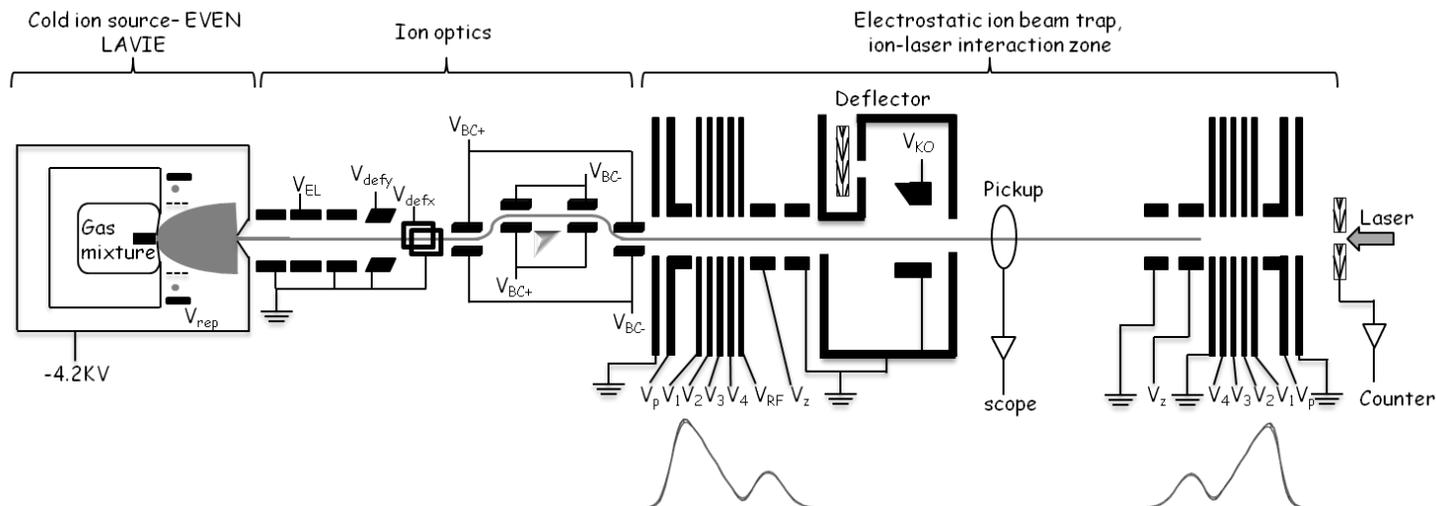


Figure 6: A schematic sketch of the experimental setup, also showing the typical potential along the optical axis of the EIBT. The ion beam induced current measured by Pickup ring, and the neutrals fragments measured by a particles detector in the end of the trap. The curve show the simulated potential that the ion beam feel by the EIBT ( $V_p, V_1, V_2, V_3, V_4, V_z$ )

## 2.1. Ion source and ion optics

$\text{SF}_6^-$  and  $(\text{SF}_6)_n^-$  molecular and cluster anions are generated in a supersonic expansion ion source located on a high voltage platform. The Even-Lavie pulsed valve releases short  $8_{\mu\text{s}}$  pulses of a specially prepared gas mixture,<sup>35</sup> typically  $\sim 420$  PSI Neon carrier gas, seeded by  $\sim 1\%$  of  $\text{SF}_6$  sample gas. The gas pulse supersonically expands into the "source" vacuum chamber that is pumped by a turbo pump to a background pressure of about  $5 \cdot 10^{-7}_{\text{Torr}}$ . A pulsed circular electron gun accelerates 200eV electrons into the expanding gas, generating a variety of cations, anions as well as metastable neutral species. The rapidly expanding carrier gas cools down to temperatures as low as 10K,<sup>35</sup> leading to cooling and even clustering of the sample molecules and ions by multiple collisions with the expanding carrier gas. The supersonic expansion is skimmed by a 3mm skimmer located 100mm downstream of the pulsed valve to select the central coldest part of the expanding gas. After the skimmer, negatively charged species are accelerated by the potential drop from the -4.2kV high voltage platform at the source toward a grounded electrode located at the entrance of ion optics chamber. Thus, the source produces a pulsed beam of anions with a kinetic energy of  $q \cdot 4.2\text{KeV}$ . As shown in figure 6, the source is followed by ion optics that include an Einzel lens as well as horizontal and vertical deflexors that focus and steer the ion beam into the EIBT with typical voltages of  $V_{\text{EL}} = 2.4\text{KV}$  with  $V_{\text{deflx}} = 28\text{V}$ ,  $V_{\text{defly}} = 30\text{V}$ . In addition, four pairs of "beam cleaner" electrodes ( $V_{\text{BC}+} = 1.032\text{KV}$ ,  $V_{\text{BC}-} = 1.038$ ) are pulsed to selectively guide ions arriving at the selected time along a chicane trajectory, while preventing neutral gas molecules as well as ion species with very different charge over mass ratio, which arrive outside the selected time window, from passing to the EIBT. Figure 7 shows a typical single pass pickup signal, recorded while grounding the EIBT electrodes, resolving the charge over mass of the anionic species that we inject into the EIBT based on their time of flight from the ion source to the pickup electrode at the center of the

trap. The most dominant peak belongs to  $SF_6^-$ , while the early shoulder is assigned to fragmentation into  $SF_n^-$  and the following peaks belong to clusters of the  $(SF_6)_n^-$  form.

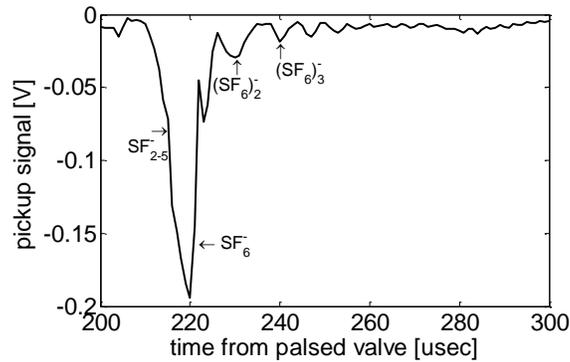


Figure 7: pickup signal measurement of a pulsed ion beam that passing through the pickup. Anions of different  $\sqrt{m/q}$  pass the pickup electrode at different times, producing negative voltage peaks as a function of time. The highest peak (220ms) is a predominantly by  $SF_6^-$ , with an unresolved tail towards low TOF composed of  $SF_n^-$  with  $n$  ranging from 2-5.  $SF_6^-$  is followed by  $(SF_6)_2^-$   $(SF_6)_3^-$  et cetera, arriving respectively at 230ms, 240ms and longer times. The negative voltage indicates negative ions, and the amplitude is proportional to the number of ions.

## 2.2. Electrostatic ion beam trap (EIBT) , RF and Kick out

The EIBT and its different modes of operation were previously described in detail in a series of manuscripts by Daniel Zajfman *and co-workers*.<sup>36-39</sup> A typical EIBT is made of two electrostatic mirrors and works similar to an optical resonator, supporting stable ion-trajectories that oscillate back and forth between two mirrors. The EIBT is placed in an ultra-high-vacuum chamber, pumped by a cryogenic pump typically to a pressure of  $10^{-10}$  Torr in order to minimize the loss of ions by collisions with the residual gas. Each electrostatic mirror is made of a series of six negative electrodes with radial concentric symmetry that defines a smooth potential. The monotonically rising potential supplied to electrodes 1 - 4 (typically:  $V_1=-1.622$ kV,  $V_2=-3.247$ kV,  $V_3=-4.873$ kV and  $V_4=-6.5$ kV) produces a reflecting potential for the 4.2keV ion beam. While the focusing is provided by an Einzel lens, typically supplied with  $V_z=3.2$ kV. The additional  $V_p$  electrode allows to either open or close the entrance mirror by setting  $V_p$  to -1.622kV or -4.42kV respectively. In order to trap a desired species,  $V_p$  is lowered to allow ions to enter the trap and is raised back once the desired species passes the pickup. For example: according to the assignments shown in figure 7,  $SF_6^-$  ions will be trapped by raising  $V_p$  230µs after the pulsed valve while raising  $V_p$  at 245µs will trap  $(SF_6)_3^-$  clusters. The ions that are inside the EIBT at the time the mirror is closed are trapped, oscillating between the mirrors. In order to synchronize the motion of specific ion species, a low RF voltage is applied to  $V_{RF}$  electrode. We typically use a 4V peak to peak sine wave at the oscillation frequency of the desired species that allows to synchronize the ion bunch to an external trigger and to keep the same ions mass from debunching due to their finite energy spread and trajectory differences.<sup>42</sup> In addition, synchronized square 100V pulses (alternating between positive and negative voltages for

technical reasons) are applied to  $V_{KO}$  during the first 100 oscillations, the deflector inside the EIBT, and effectively “kick out” the majority of the ions oscillating out of synch with the desired ion species oscillating at the selected RF frequency<sup>40</sup>. Thus, we obtain a trapped ion bunch mass selected and synchronized to the laser pulse.

### 2.3. Laser

We are using an EKSPLA tunable wavelength Laser base on optical parametric oscillator (OPO) pumped by a ND-YAG laser.<sup>41</sup> At a repetition rate of 10Hz, the OPO provides  $\sim 4$ ns pulses with tunable wavelength. The OPO signal wavelength is computer controlled in the range of: 420 – 709nm with line width of  $< 5\text{cm}^{-1}$  (0.62meV) and scanning step of 0.1nm. The laser beam is further shaped and directed towards the EIBT by a telescope and a set of 3 mirrors that allow passing the laser beam through the 3mm hole in the MCP detector and overlapping it with the optical axis of the EIBT. Figure 8a and 8b show a typical measured output power of the OPO signal and idler as a function of photon energy. Only the OPO signal is used in the experiments described here, due to a combination of vanishing PDE and low laser power below 2 eV photon energies.

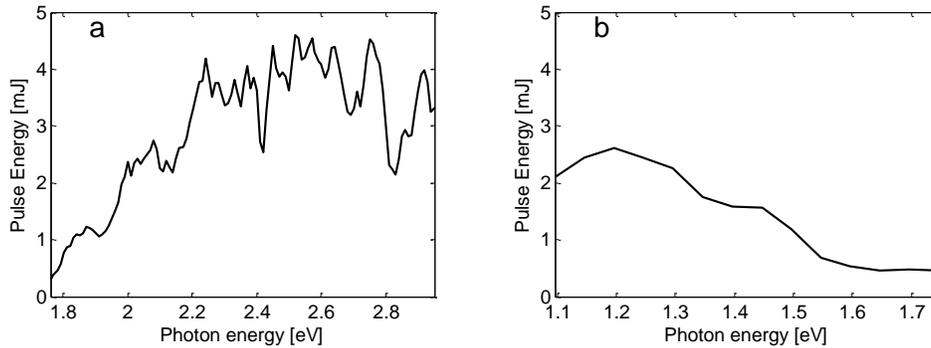


Figure 8 a,b: typical laser pulse energy output from the OPO signal(a) and idler(b).

### 2.4. Timing and synchronizations

Figures 9a and 9b describe the timing scheme of a typical experiment, showing the sequence of computer controlled triggers supplied to the experimental apparatus. The experiment is synchronized to the laser pulse timing. The first timing signal triggers the pulsed valve, at  $t=0$  which is  $\sim 75$ ms after the previous laser pulse. The delay from the laser pulse is calculated such that the next laser pulses (arriving at 25ms, 125ms and 225ms) will hit the ion bunch when it is at the center of the trap and is moving towards the exit mirror, such that neutral photodetachment products that retain the parent ion velocity move toward the MCP detector. The pulsed valve trigger is followed by a  $10\mu\text{s}$  ionizer pulse, delayed by  $69\mu\text{s}$ . The beam cleaner is typically activated after  $208\mu\text{s}$  for  $8\mu\text{s}$ . As shown in figure 7,  $\text{SF}_6^-$  anions reach the center of the

EIBT at about 220 $\mu$ s and the EIBT entrance mirror is closed by switching the  $V_p$  at 230 $\mu$ s and keeping it high until the end of the cycle at which point remaining ions are released from the trap. Once the trap is closed, a synchronized sine wave is applied to the RF electrode. The periodic “kick out” pulse sequence applied to the  $V_{KO}$  deflector electrode for 100 oscillations only and is delayed by a phase of  $1.4\pi$  with respect to the RF sine wave. The phase shift is optimized for keeping the desired ion bunch, while kicking out all other species with a  $\frac{\Delta m}{m} = 7 \cdot 10^{-3}$  mass resolution<sup>40</sup>. It is important to note that it is necessary to match the RF frequency to be an integer times the exact 9.9986Hz repetition frequency of the laser in order to synchronize the ion bunch with more than one laser pulse.

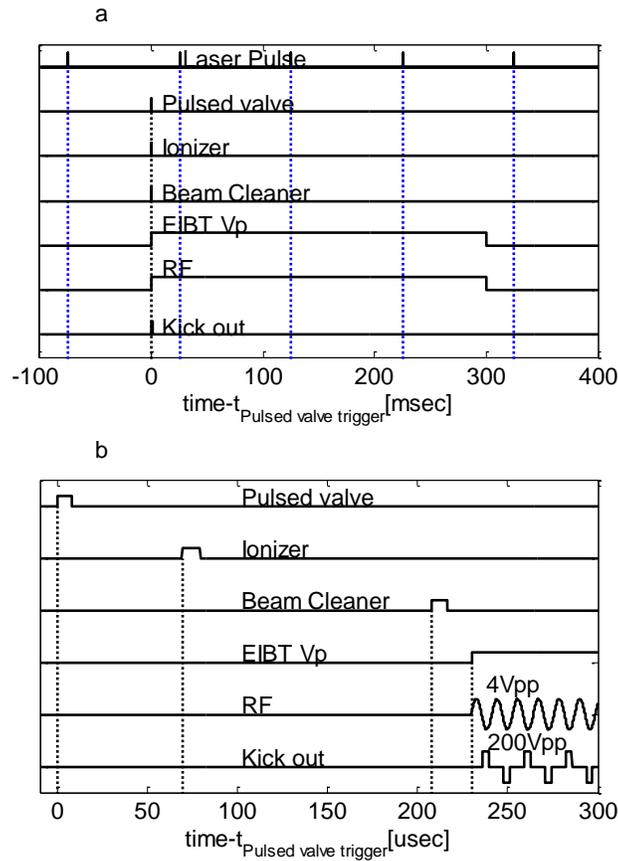


Figure 9 : Experimental timing scheme for  $SF_6^-$ : a. show synchronization between the laser to the setup triggers, after the first laser pulse the pulse valve starts the sequence, and that the laser hits three times (blue dots) at the trapped ions, while the trapped ions are synchronized toward the exit mirror. The Kickout is working for  $\sim 1$ ms, so it's too thin to see its thickness here. b. is a zoom of a to the pulsed valve area to see the different triggers. First a gas re-enters from the pulsed valve for 8 $\mu$ s, second the gas gets ionized for 10 $\mu$ s, third the ion beam gets clean from different masses, energies, and neutrals by the Beam Cleaner, and finally trapped inside the EIBT by rising  $V_p$ , gets into resonance by the RF, and mass selected to the desired ion by the Kickout.

### 3. Data acquisition

As shown in figure 6, the ions and their neutral products can be detected by three separate detectors: a multi channel plate (MCP)<sup>43,44</sup> that records the impact of neutral products of both ion collisions with the residual gas and neutral products of ion interaction with the laser pulse. The pickup ring electrode that acts as an antenna, measuring the induced image charge of the ion beam inside the EIBT.<sup>38,45</sup> A specially designed compact deflector that I designed, constructed and installed inside the EIBT, which will allow to select product ions according to their charge over mass and detect them on a dedicated micro sphere plate detector (MSP)<sup>46</sup>. The first two detection schemes were previously explained in detail in<sup>38,45</sup>, while the compact deflector was designed and built as a part of my M.Sc. work and will be described in more detail here.

#### 3.1. Neutral product detection

Neutral products of the ion beam interaction with residual gas collisions, blackbody radiation<sup>47</sup> and laser pulses are not deflected by the EIBT mirror. Neutrals that fly out of the exit mirror can hit the MCP detector located 67cm after the trap (see figure 6). Although such neutral products have no charge, their high kinetic energy is sufficient to produce a signal on the MCP detector. The small current peaks on the MCP anode are amplified and then discriminated by a constant function discriminator (CFD) producing a digital NIM timing signal, which is converted to a TTL signal and fed into a "NI-PCI-6602 DAQ" PC card that records the neutral products arrival times to the detector with 100ns time resolution, limited by the maximal counter clock frequency. In order to avoid saturation of the acquisition we limit the maximal count rates to 1 over 100ns by reducing the number of trapped ions and beginning the measurement at 15ms trap time.

#### 3.2. Capacitive pickup ion detection

The time evolution of the ion bunch size and shape is directly and non-destructively measured by a sensitive antenna ring electrode (pickup) that is slightly shifted from the center of the trap towards the exit mirror<sup>42</sup>. The pickup is connected to an amplifier and then recorded on a dedicated scope that records the pickup signal during an entire trapping cycle with sampling frequency of 1MHz. The pickup electrode length is 7 mm and has an inner diameter of 18 mm. The total capacitance of the pick-up electrode, including its support rod and vacuum feed-through is  $C \sim 10\text{pF}$ . Due to various slow potential drifts and noises in the lab it is not possible to determine precisely the induced DC charge from the measured voltage. However, the AC image charge induced by the passage of an ion bunch can be determined precisely as will be further explained in the data analysis section 2.2. In order to insure that the ions are all bunched an RF voltage of 4V is applied to the  $V_{\text{RF}}$  electrode of the trap, effectively bunching the trapped ions<sup>42</sup> allowing us to neglect any DC contribution. We can therefore relate the number of ions in the trap to the area of a pulse produced by the ion bunch as it passes through the pickup.<sup>42</sup>

$$(7) \quad N_{ions} = \frac{C}{L_p} \frac{v}{eq} A$$

Where  $L_p$  is the pickup length (7mm), and  $C$  in the capacitance ( $\sim 10\text{pF}$ ),  $v$  is the ion velocity and  $A$  is the area under the oscillation. For example, considering the  $\text{SF}_6^-$  peak shown in figure 7,  $A$  is about  $0.1\text{ mV}\cdot\text{sec}$ ,  $v$  is about  $70\text{ km/sec}$ , resulting in about  $7\cdot 10^8$  ions in the bunch.

### 3.3. Ion fragment spectrometer – compact deflector setup

So far the setup allowed detection of the trapped ions and its neutral products, but lack detection of charged ionic products that are blocked by the exit mirror of the trap and cannot reach the MCP detector outside the EIBT. The compact deflector setup is designed to fit inside the EIBT and allow to separate product ions by their kinetic energy. The kinetic energy of a daughter ion can be directly related to its mass by  $m_{\text{daughter}}/m_{\text{parent}}\cdot 4.2\text{keV}$ , as the kinetic energy released in the dissociation event can be generally neglected compared to the kinetic energy of the beam in the lab frame. With this ability it can resolve daughter ions and measure the branching ratio between the possible charged products. Ion trajectory simulations presented in figure 10 show the trajectory of a trapped ion beam that is deflected with a pulsed voltage applied to the upper deflector electrode ( $V_{\text{def}}$ ) after  $200\mu\text{sec}$  trapping time. The deflection angel depends on the kinetic energy, the charge of the ion and the applied voltage on the upper electrode. In this simulation, a  $2150\text{V}$  potential is applied to  $V_{\text{def}}$ , directing the  $4.2\text{ keV}$  ions towards the detector through a narrow  $2\text{mm}$  slit. In order to detect daughter ions that have a fraction of the  $4.2\text{ keV}$  kinetic energy of the parent ion, a corresponding lower  $V_{\text{def}}$  is applied.

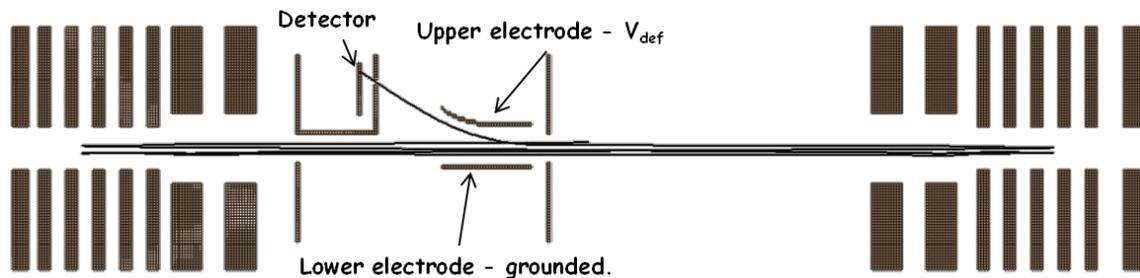


Figure 10: a simion simulation of  $\text{SF}_6^-$  with kinetic energy of  $4.2\text{KeV}$  trapped inside the EIBT for  $200\mu\text{sec}$ , then  $V_{\text{def}}$  is changed from  $0$  to  $2150\text{V}$  and deflect the ion toward the detector.

Figure 11 shows the simulated correlation between the kinetic energy of ions passing the  $2\text{mm}$  slit and the applied  $V_{\text{def}}$ . Such simulations allowed us to optimize the slit width to obtain the maximal detection efficiency with acceptable daughter ion mass resolution. Figure 12 shows a realistic simulation of detector response as a function of applied potential considering  $\text{SF}_6^-$  parent ion signal and  $\text{SF}_5^-$  and  $\text{SF}_4^-$  daughter ions. The simulation includes an estimated  $100\text{eV}$  FWHM Gaussian energy spread of the parent ion beam, a  $100\text{meV}$  kinetic energy release in the dissociation process, nevertheless simulation shows that the daughter ions can be clearly mass resolved. Figure 13 shows the mechanical design of the compact deflector setup and its photograph before it was installed in the EIBT system.

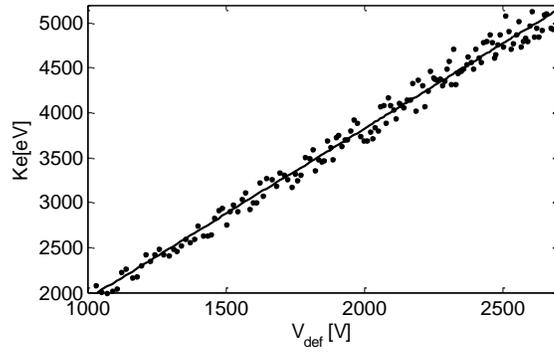


Figure 11: simion simulation that simulated ions that hit the detector, for different energies, with a fit to linear equation result slope of 1.9-Cb.

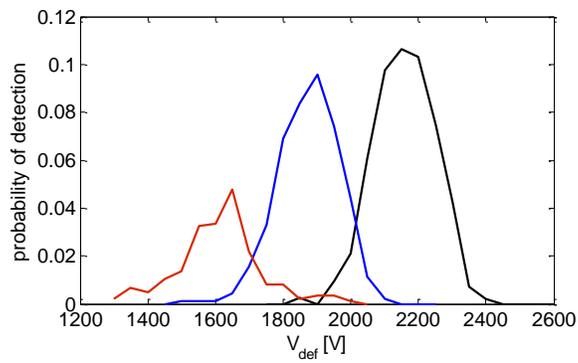


Figure 12: Simulated detection probabilities of  $SF_6^-$  (solid black) and its photodissociated daughter fragments of  $SF_4^-$  (solid red) and  $SF_5^-$  (solid blue) as a function of the upper electrode voltage.

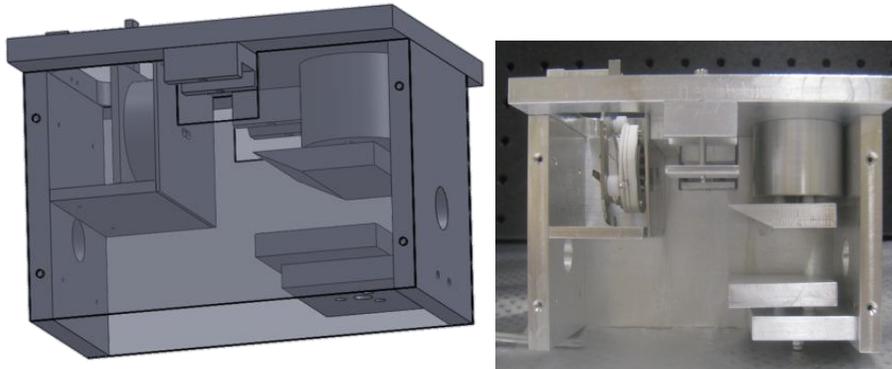


Figure 13 : final design of the deflector (left) and on the deflector before it been installed (right).

#### 4. Data analysis:

In the data analysis section I explain how the raw data collected from the pickup setup and the MCP setup is analyzed to derive the photodetachment efficiency (PDE) curve for a selected ion species as a function of photon energy. In the low flux regime, PDE can be derived according to the following equation:

$$(8) \quad PDE = \frac{1}{F} \frac{N_{product}}{N_{ions} \cdot \Phi}$$

Where  $N_{products}$  represents the number of neutral photodetachment products,  $N_{ions}$  represents the number of anions and  $\Phi$  is the photon flux and  $F$  is one over the geometric overlap of the ion bunch and the laser beam.

##### 4.1. Measurement of the number of neutral photo-products

Figure 14 shows the MCP signal not in coincidence with a laser pulse, averaged about  $2 \cdot 10^5$  measurements of trapped  $SF_6^-$  anions. The signal ( $N_{mcp}$ ) is predominantly due to neutralization by residual gas collisions and can be related  $N_{ions}$  to the number of trapped ions by:

$$(9) \quad N_{mcp}(t) = \eta k N_{ions}(t) dt$$

Where  $N_{mcp}$  is the number of hits on the MCP in a  $dt$  long time window,  $\eta$  is the detection efficiency of the detector and  $k$  is the neutralization rate constant. As will be explained in the following, these average hit rates that are not in coincidence with the laser are subtracted as background to the photodetachment process. As can be clearly seen the decay has more than one component and therefore  $k$  is in fact a function of time, making it difficult to conclude on the absolute number of trapped ions from a measurement of the neutral yield alone.

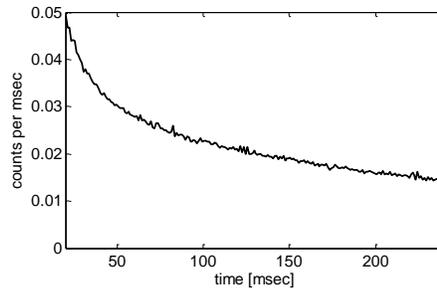


Figure 14 : life time of the neutrals fragments due to residual gas collision detect by MCP

Figure 15 shows a zoomed in view around the time of the laser pulse. The sharp peak at  $t=0$  is due to electronic noise arriving in coincidence with the firing of the laser pulse. For  $SF_6^-$  parent ions, neutral photo detachment products born at the center of the trap are expected to arrive in a time window between  $8\mu s$  and  $11.8\mu s$  after the laser pulse due to the time of flight (TOF) from EIBT to the MCP. Naturally the time of flight scales with the square root of the parent ion mass for other species, allowing further cleaning of the PDE spectra from possible contaminations of

the ion sample in the data analysis stage. The small tail visible in the log scale representation in figure 15 is due to a negligible number of neutralization events that occur not in the field free region of the trap but in the mirror region that nevertheless arrive to the MCP and produce a measurable signal, albeit arriving with lower speed and at therefore at later times. The periodic background, which is 0.001 times smaller than the photodetachment signal, is due to residual gas collisions and is modulated due to the oscillations of the ion bunch.

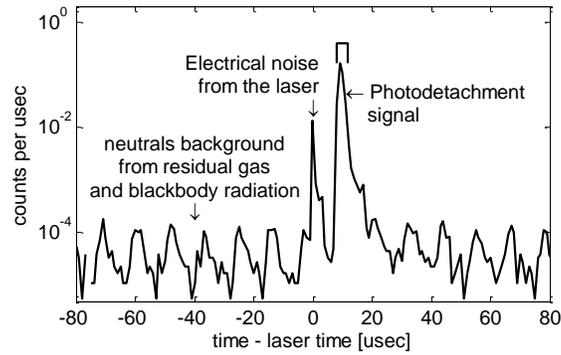


Figure 15: a zoom around the laser time of the raw data average of about  $2 \cdot 10^5$  measurements. When after synchronization, the ions-laser photodetachment neutral signal (square marker) comes with the residual gas neutral signal.

Assuming that the decay of the ion beam occurs on long time scales we calculate the average background signal in one oscillation period,  $N_{BG}(t)$ , using oscillations -26 to -5 and 5 to 26 relative to the laser pulse. The periodic average background is then subtracted from the MCP signal as shown in Figure 16, with a signal to noise level of about  $>10^4$ .

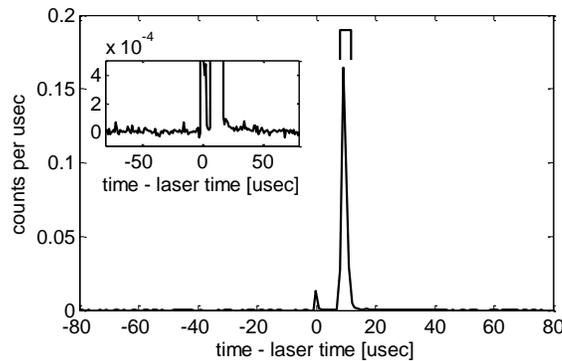


Figure 16: The MCP setup signal around the laser time after a background subtraction, when the photodetachment signal mark with black square, while there is still electrical noise from the laser at zero time. The tale of the signal is neglect to get a better signal to noise ratio.

Since our MCP has a finite size of 40mm diameter with a dead area of 7mm diameter due to the hole in its center, the finite detection efficiency of the detector needs to be evaluated. Following Atia *et al*<sup>48</sup> we rely on a detailed ion trajectory simulation to calculate the acceptance of the

detector. Figure 17 shows the simulated distribution of particle hits on the detector as a function of radial distance. Taking into account the dead area at the center of the MCP we calculate that the geometric acceptance of the detector is 63%. When evaluating the detector efficiency we need also to consider the typical ~58% open area efficiency of the MCP detector<sup>49</sup> to get an overall  $\eta \approx 0.365$ . We can now evaluate the number of neutral products due to the laser as

$$(10) \quad N_{product} = (N_{MCP} - N_{BG}) / \eta$$

Where  $N_{MCP}$  and  $N_{BG}$  are the total number of counts arriving at the above time window, respectively in and out of coincidence with the laser.

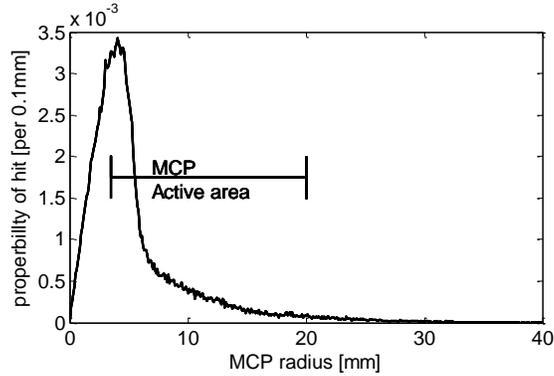


Figure 17: a simion simulation of the active area of the MCP

#### 4.2. Measurement of the number of ions in the trap ( $N_{ions}$ )

Figure 18a shows a typical raw pickup data averaged over 31 injections, including signal from the  $SF_6^-$  ion bunch oscillations as well as a lot of electronic noise. Figure 18a inset shows a zoomed in view of the periodic passes of the ion bunch trough the pick-up electrode. The Fourier transform of the raw data, shown in Figure 18b, clearly shows a strong signal at the frequency of 87062Hz and at its harmonics, corresponding to mass 146amu of  $SF_6^-$ .

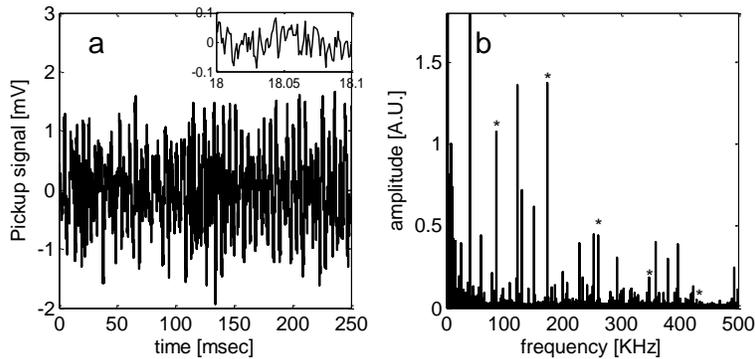


Figure 18: a. raw signal of 31 injections average of the pickup, with phase lock by the RF, when the signal is with a lot of electric noise, the ion beam signal is less then 1mV, in the inset it is able to see a 11.7 $\mu$ s periodic signal. b. show a raw Fourier transform of the signal, it's able to see the ion frequency at 87KHz and its harmonics, as well as a electric noise.

Figure 19a shows the FFT spectrum after application of a filter that passes only the desired frequency and its harmonics. Figure 19a inset shows the Lorentzian shape of the applied filter at the fundamental frequency with a FWHM of 870Hz, selected to preserve bunch dynamics on the few ms and longer time scales that should not affect the sought after time evolution of the number of ions. Intuitively, since the ion bunch pass through the pickup two times each oscillation, one can expect the even harmonics (174KHz, 348KHz) to be stronger than the odd harmonics (87KHz, 261KHz, 435KHz). In fact, following reference<sup>42</sup> we can derive an analytical expression for the harmonics spectrum, assuming Gaussian bunch shapes of width  $\Gamma$ , by performing a Fourier transform on the periodic signal  $S(t)$ :

$$(11) \quad S(t) = A(t) \sum_{j=-\infty}^{\infty} \frac{1}{\sqrt{2\pi}\Gamma} \left( \exp\left(-\frac{(t-jT_0)^2}{2\Gamma^2}\right) + \exp\left(-\frac{(t-(j+\kappa)T_0)^2}{2\Gamma^2}\right) \right)$$

Where  $A(t)$  is the time dependent area of each temporal peak corresponding to the decaying number of ions,  $j$  is the oscillation number and  $T_0$  is the oscillation period. The expressions show two peaks for each oscillation: a forward and a backward pass through the pick-up with  $\kappa$  representing the relative location of the pickup in the trap. In our setup the pick-up is slightly removed from the center of the trap toward the exit mirror, corresponding to  $\kappa=0.4511$ . Equation 12 shows the derived peak intensity of each harmonic.

$$(12) \quad S_n \propto \sqrt{2 + 2\cos(2\pi n\kappa)} \cdot \exp\left(-\frac{(n\Gamma/T_0)^2}{2}\right)$$

Where  $n$  is the harmonic number, and  $S_n$  is the harmonic amplitude. This equation is in agreement with the observed FFT spectrum from figure 18b, where the second harmonic is stronger than first harmonic. Figure 19b shows a zoom on typical filtered pickup signal obtained by inverse Fourier transform. Note that since the image charge is of negative ions, the minima correspond to the ion bunch passing through the pickup and the maxima are when the ion beam is in one of the mirrors. The short and long half oscillations are clearly visible, since the pickup is shifted slightly toward the exit mirror we can assign the times at which the ions are in the field free region moving towards the exit mirror. To capture the evolution of the number of ions as a function of time, we subtract the minima from the maxima signal, averaged over every 200 oscillations and multiply it by  $\sqrt{2\pi}\Gamma$  to evaluate the characteristic bunch area as a function of trapping time,  $A(t)$ . We find that that for our experimental conditions Gamma scales with the oscillation time and was equal to  $\Gamma=T_0/12.5$ . Using eq. 7 we can now evaluate the number of trapped ions as a function of time. Figure 20 shows the number ions for a typical trapping of  $SF_6^-$ , we find that a fit of one exponent can describe good the data, that help us to interpolate the number of ions in the EIBT exactly at the time of laser shot.

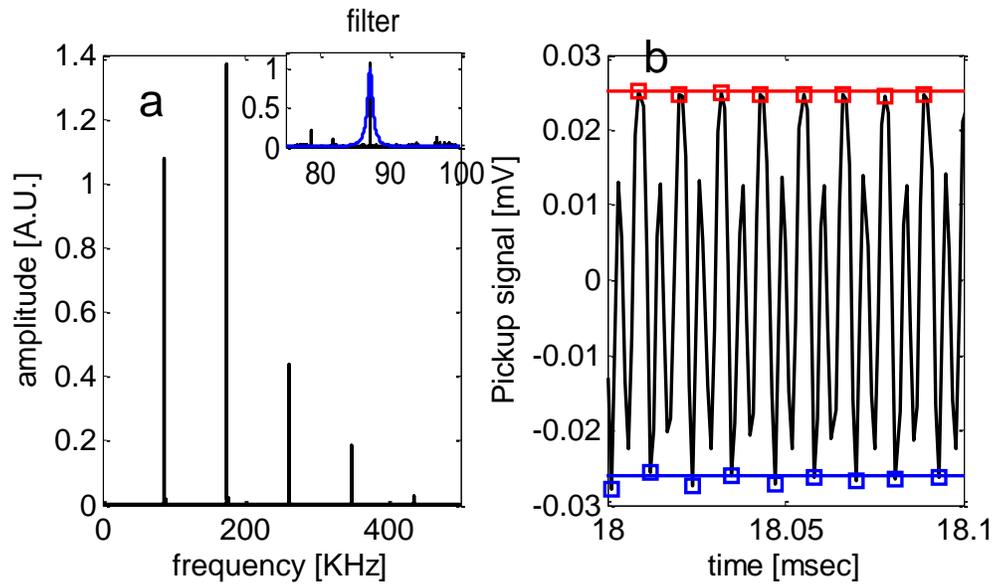


Figure 19 a. a 31 injection average of a clean Fourier signal after Lorentzian filtered, it is able to do that due to the phase lock by the RF. the inset show a zoom on the raw Fourier signal (black line), and the Lorentzian filter (blue line). b. a clean pickup time signal (solid black line), it is easy to see the oscillation of the ion beam. In this figure there is a several element in this figure, 1: The pickup measure the image charge of an anion beam so the min points are the picks of when the ion beam pass inside the pickup and the max point are when the ion beam inside the mirror, 2: The ion beam pass two times through the pickup in each oscillation, therefore there is two picks in every oscillation, a low pick and a high pick, the low pick is when the ion beam is in the entrance mirror side which is farther away from the pickup, the high pick is when the ion beam is in the exit mirror side which is closer, and if the ion beam length is big enough the pickup can also feel some of the ion beam of the next oscillation. Every 200 max and min points of every 1-2 oscillations (red and blue squared respectively) had been chosen to carried out smooth and spline (solid red and blue lines respectively), the differential between these two lines is proportional the number of ions in the EIBT.

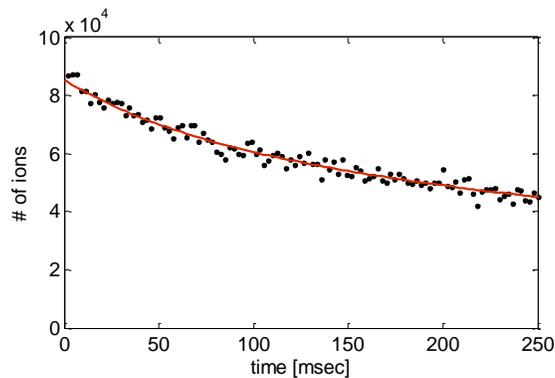


Figure 20: The number of ions in the EIBT calculated by eq. 7 (black dots), with a fit of one exponent (red line) that result the lifetime  $\tau_1=68\text{ms}$ .

#### 4.3. Photo detachment efficiency – PDE

The photon flux  $\Phi$  is calculate from the measured laser power by dividing the energy per pulse by the photon energy and the evaluated  $A_{laser} \sim 7.1 \text{ mm}^2$  area of the laser beam

$$(13) \quad \Phi = \frac{\text{laser pulse energy}}{h\nu \cdot A_{laser}}$$

The same ion trajectory simulation described in the previous section was used to estimate the geometric overlap of the ion bunch ( $\sim 3 \text{ mm}$  beam radius) and the laser beam ( $\sim 1.5 \text{ mm}$  radius), resulting in a correction factor of  $F=1/4$  described in eq. 8. Combining the number of ions, neutral products and laser flux according to equation 8, figure 21 presents a typical PDE curve of  $\text{SF}_6^-$ .

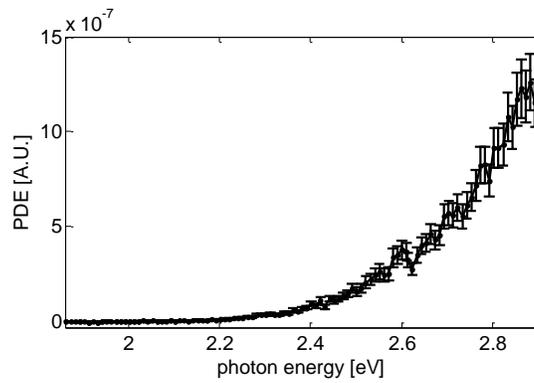


Figure 21:  $\text{SF}_6^-$  PDE (black dots)

## 5. Results and discussion

Although ions are formed in a cold ion source, it is still valuable to check for significant changes in the measured PDE curve as a function of trapping time in the room temperature trap environment. Figure 22a shows a comparison of the  $\text{SF}_6^-$  PDE curve shape for three different laser shots, probing the ions at 25,125 and 225 ms after the ions enter the trap. The three spectra are normalized according to the area under the PDE curve in order to directly compare the shape of the curves. As seen in the figure the structures of the PDE of all three shots look the same within the error bars, which indicates that the PDE of  $\text{SF}_6^-$  does not change as a function of slow temperature changes during our experiment. In figures 22 a and b we compare the PDE curves recorded at different trapping times for  $(\text{SF}_6)_2^-$  and  $(\text{SF}_6)_3^-$  respectively, reaching similar conclusions.

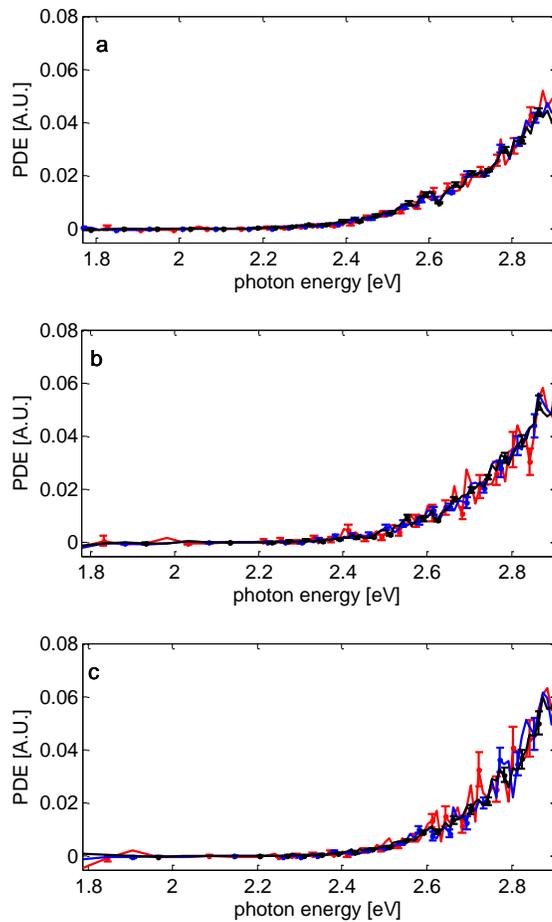


Figure 22 : comparing between the PDEs of  $\text{SF}_6^-$ ,  $(\text{SF}_6)_2^-$  and  $(\text{SF}_6)_3^-$  in three different trapping time, 25ms(black), 125ms(blue), 225ms (red) a,b,c respectively.

Combining the data from all three laser shots to obtain better signal to noise ratios we can now compare the PDE curves of  $\text{SF}_6^-$  to the  $(\text{SF}_6)_2^-$  and  $(\text{SF}_6)_3^-$ . Figure 23 shows that within the error

bars there is also no big changes in the shape of the PDE curve as one or two neutral SF<sub>6</sub> molecules are added to the SF<sub>6</sub><sup>-</sup> anion.

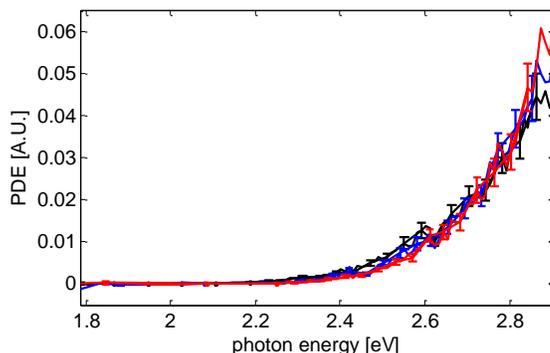


Figure 23: comparing between the PDEs of SF<sub>6</sub><sup>-</sup>, (SF<sub>6</sub>)<sub>2</sub><sup>-</sup> and (SF<sub>6</sub>)<sub>3</sub><sup>-</sup> (black, blue, red respectively)

To better compare the PDE we use threshold power law function to fit the PDE curve.<sup>50</sup>

To describe the results we use a fit to the function

$$(14) \quad y = a \cdot \theta(E - E_0) \cdot (E - E_0)^n$$

Where  $\theta$  is a Heaviside function and the free fitting parameters are a scaling factor “ $a$ ”, the power  $n$  and the threshold energy  $E_0$ . Figure 24 shows SF<sub>6</sub><sup>-</sup> PDE and the fit for the data points of equation 14, for which we obtain fitted power  $n=4.5\pm 0.8$  and thresholds  $E_0= 1.8\pm 0.1$ eV, for (SF<sub>6</sub>)<sub>2</sub><sup>-</sup> and (SF<sub>6</sub>)<sub>3</sub><sup>-</sup> we obtain similar values. When applying this fit to our harmonic oscillator PDE model calculation shown in fig. 5 and described in the background section, with the anion equilibrium bond length of 1.70Å,  $EA_{ad}=1$ eV and including SNR comparable to the experimental conditions, we obtained fitted power  $n=4.4\pm 0.5$  and  $E_0=1.70\pm 0.1$ eV. Note that fitted  $E_0$  parameter is not directly indicative of  $EA_{ad}$ , even in the modulated PDE.

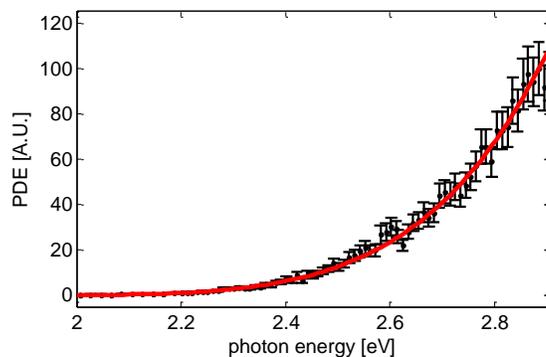


Figure 24: SF<sub>6</sub><sup>-</sup> PDE (black) and a fit of equation 14 (red)

Although we expected that adding additional SF<sub>6</sub> molecules to the SF<sub>6</sub><sup>-</sup> anion might delocalize the charge and make the anion effective equilibrium bond length smaller, in between the equilibrium bond lengths of SF<sub>6</sub><sup>-</sup> anion to SF<sub>6</sub>. As was shown in fig 5 that would result in a higher PDE for the low energies tail for the cluster anions compare to SF<sub>6</sub><sup>-</sup>. Figure 25 shows the PDE curves of SF<sub>6</sub><sup>-</sup>, (SF<sub>6</sub>)<sub>2</sub><sup>-</sup> and (SF<sub>6</sub>)<sub>3</sub><sup>-</sup> and the equation 14 fit to SF<sub>6</sub><sup>-</sup> in cyan, we find out that SF<sub>6</sub><sup>-</sup>

exhibits a higher tail toward the low photon energies. Figure 26 showing that by calculating the harmonic oscillator PDE model with equilibrium bond length of  $1.70\text{\AA}$ ,  $EA_{ad}=1\text{eV}$  in zero temperature comparing to room temperature find out that this kind of tail can come from vibrationally hot molecules. Therefore, we assign this small change to relatively hot  $\text{SF}_6^-$  molecules that can be produced in the low pressure parts of the supersonic expansion due to insufficient cooling. On the other hand, weakly bound dimers and trimers cannot be formed at elevated temperature due to their low binding energies and therefore strictly cold. We conclude that when adding another  $\text{SF}_6$  molecules to  $\text{SF}_6^-$  the charge stayed localized on the  $\text{SF}_6^-$  anion.

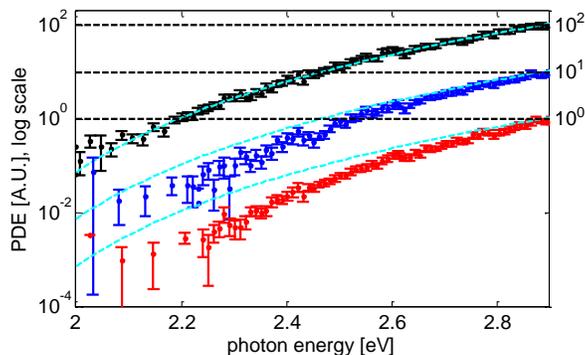


Figure 25: normalized PDE's of  $\text{SF}_6^-$ ,  $(\text{SF}_6)_2^-$  and  $(\text{SF}_6)_3^-$  to 1, 10, 100 respectively (black, blue, red respectively), and the fit of equation 14 for  $\text{SF}_6^-$  (cyan), comparing to each molecule. While the shape of  $(\text{SF}_6)_2^-$  and  $(\text{SF}_6)_3^-$  looks similar,  $\text{SF}_6^-$  have a tail toward the low photon energy, we assume that it come from part of  $\text{SF}_6^-$  anion created in the source were hot, while  $(\text{SF}_6)_2^-$  and  $(\text{SF}_6)_3^-$  are cold.

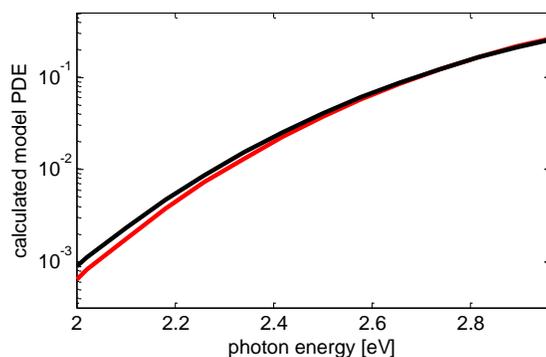


Figure 26: numerical calculated PDE of harmonic oscillator with equilibrium bond length of  $1.70\text{\AA}$ ,  $EA_{ad}=1\text{eV}$  in zero temperature (red) comparing to room temperature (black)

As we previously explained, when considering only a direct detachment mechanism, the PDE curve should be similar to the sum over the PES spectrum. Figure 27 shows the measured PDE of  $\text{SF}_6^-$  in our lab compared to an integral over the  $\text{SF}_6^-$  photoelectron spectrum adopted from Bopp *et al*<sup>24</sup>. We observe that within the error bars there is a very good agreement with no observable indication of contribution from indirect detachment.

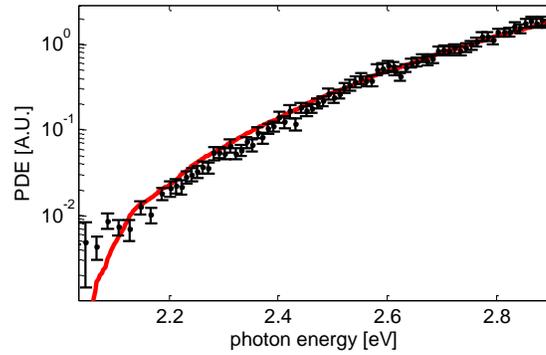


Figure 27: comparing between the  $\text{SF}_6^-$  PDE (black dots) and the integral on  $\text{SF}_6^-$  photoelectron spectrum adopted from Bopp *et al.*<sup>24</sup> (solid red)

After comparing the shape, using equation 8 we can compare the absolute PDE of different ions. Figure 28 compare between the absolute PDE's of  $\text{SF}_6^-$  and  $(\text{SF}_6)_2^-$  (black, red respectively), since within the error bar they have the same PDE, we conclude that the addition  $\text{SF}_6$  molecule in  $(\text{SF}_6)_2^-$  doesn't change the PDE magnitude as well as its shape. On the other hand, when we evaluate the residual gas neutralization rate constant " $K$ ", by dividing  $N_{mcp}$  by  $N_{ions}$  according to equation 9, we get  $1.5 \cdot 10^{-3}$  and  $5 \cdot 10^{-3}$  for  $\text{SF}_6^-$  and  $(\text{SF}_6)_2^-$  respectively at  $t=35\text{ms}$ . This emphasizes the point that photodetachment cross section does not seem to change when adding another  $\text{SF}_6$  molecule to the cluster, while collision induced neutralization increase by a factor of 3.33.

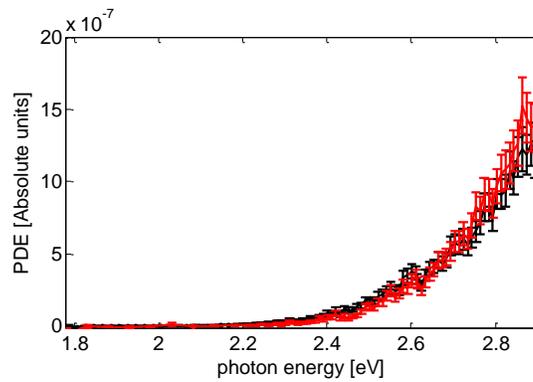


Figure 28: comparing the PDE's of  $\text{SF}_6^-$  and  $(\text{SF}_6)_2^-$  (black, red respectively).

Figure 29 Show us absolute photodetachment of  $\text{SF}_6^-$  adopted from the work of Mock *et al.*<sup>30</sup> Compared to our  $\text{SF}_6^-$  measurement. Our measurement extended the range of measured photon energies down to 1.8eV, measuring signals that are few orders of magnitude below the detection limit.<sup>30</sup> shown by the dashed line. This was possible due to the use of neutral product detection scheme for efficient fast ion beam technique.

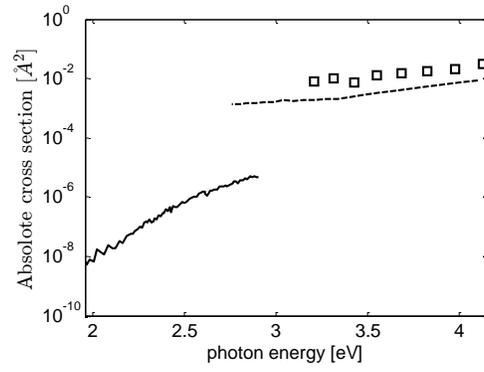


Figure 29:  $\text{SF}_6^-$  absolute cross section (black squares) and the detection limit (dashed line) from the work of Mock *et al.*<sup>30</sup> completed with  $\text{SF}_6^-$  PDE measurement.



## 6. Summery and Conclusions

The photodetachment efficiency curves of  $\text{SF}_6^-$ ,  $(\text{SF}_6)_2^-$  and  $(\text{SF}_6)_3^-$  have been measured.

We found that within the error bar all the measurements of the three species have the same shape, except of a small higher tail under 2.3eV for  $\text{SF}_6^-$  that come from relativity hot ions. We also found that the all the PDE curve shapes are independent of storage time for our experiment time range. We also found the absolute PDE magnitude is the same for  $\text{SF}_6^-$  and  $(\text{SF}_6)_2^-$ . The good agreement between the integral of privies PES measurement and our PDE curve shape indicate there is no significant contribution from indirect transition as equation 5 suggested. We conclude that the structure of  $\text{SF}_6^-$ ,  $(\text{SF}_6)_2^-$  and  $(\text{SF}_6)_3^-$  PDEs are dominated by the  $\text{SF}_6^-$  anion where the additions  $\text{SF}_6$  molecules don't change the PDE. We also design, construct and installed a photo fragment spectrometer that future experiments will use for resolving partial photodissociation cross sections as a function of photon energy.

## 7. Reference

1. P.G. Ashbaugh, D.W. McAdam, and M. F. J. SF<sub>6</sub> - ITS Properties AND USE as a GASEOUS INSULATOR IN VAN DE GRAAFF ACCELERATORS. *IEEE Trans. Nucl. Sci.* 266 (1965).
2. Newell, W. M. J. and W. R. Absolute elastic differential cross sections for electron scattering from SF<sub>6</sub>. *J. Phys. E At. Mol. Opt. Phys.* 24, 473–487 (1991).
3. Qureshi, N. H. Mi. and A. H. A REVIEW OF ELECTRICAL BREAKDOWN IN MIXTURES OF SF<sub>6</sub> AND OTHER GASES. *IEEE Trans. Electr. Insul* EI-14, 1 (1979).
4. Hussain, S. Q. *et al.* Light trapping scheme of ICP-RIE glass texturing by SF<sub>6</sub>/Ar plasma for high haze ratio. *Vacuum* 94, 87–91 (2013).
5. Gharoooni, M. *et al.* A novel non-sequential hydrogen-pulsed deep reactive ion etching of silicon. *J. Micromechanics Microengineering* 23, 095014 (2013).
6. Cheng, J.-H., Bartos, S. C., Lee, W. M., Li, S.-N. & Lu, J. (Ching-H. SF<sub>6</sub> usage and emission trends in the TFT-LCD industry. *Int. J. Greenh. Gas Control* 17, 106–110 (2013).
7. Ko, M. K. W. *et al.* Atmospheric Sulfur Hexafluoride ' Sources , Sinks and Greenhouse Warming. 98, (1993).
8. Reddmann, T., Ruhnke, R. & Kouker, W. Three-dimensional model simulations of SF<sub>6</sub> with mesospheric chemistry. *J. Geophys. Res.* 106, 14525 (2001).
9. Houghton, et al. GWPs and atmospheric lifetimes of halocarbons and other halogenated compounds. *IPCC SAR* (1996). at <http://www.ipcc.ch/ipccreports/sres/emission/index.php?idp=123>
10. Rigby, M. *et al.* History of atmospheric SF<sub>6</sub> from 1973 to 2008. *Atmos. Chem. Phys.* 10, 10305–10320 (2010).
11. Brenninkmeijer, C. A. M. Atmospheric SF<sub>6</sub> : Trends , Sources , and Prospects. *Environ. Sci. Technol* 32, 3077–3086 (1998).
12. Troe, J., Miller, T. M. & Viggiano, A. A. Communication : Revised electron affinity of SF<sub>6</sub> from kinetic data Communication : Revised electron affinity of SF<sub>6</sub> from kinetic data. 1. *Troe J, Mill. TM, Viggiano AA. Commun. Revis. electron Affin. SF6 from Kinet. data Commun. Revis. electron Affin. SF6 from Kinet. data. 2012;1211021–4.* 136, 121102 (2012).
13. Compton, R. N., Reinhardt, P. W. & Cooper, C. D. Collisional ionization between fast alkali atoms and selected hexafluoride molecules. *J. Chem. Phys.* 68, 2023 (1978).
14. Streit, G. E. Negative ion chemistry and the electron affinity of SF<sub>6</sub>. *J. Chem. Phys.* 77, 826 (1982).

15. Chen, E. C. M., Wiley, J. R., Batten, C. F. & Wentworth, W. E. Determination of the electron affinities of molecules using negative ion mass spectrometry. *J. Phys. Chem.* 98, 88–94 (1994).
16. Viggiano, A. A., Miller, T. M., Friedman, J. F. & Troe, J. Low-energy electron attachment to SF<sub>6</sub>. III. From thermal detachment to the electron affinity of SF<sub>6</sub><sup>-</sup>. *J. Chem. Phys.* 127, 244305 (2007).
17. Eisfeld, W. Highly accurate determination of the electron affinity of SF<sub>6</sub> and analysis of structure and photodetachment spectrum of SF<sub>6</sub><sup>-</sup>. *J. Chem. Phys.* 134, 054303 (2011).
18. Gerchikov, L. G. & Gribakin, G. F. Electron attachment to SF<sub>6</sub> and lifetimes of SF<sub>6</sub><sup>-</sup> negative ions. *Phys. Rev. A* 77, 042724 (2008).
19. Appelhans, A. D. & Delmore, J. E. Refinement of the autoneutralization lifetimes of short lived states of SF<sub>6</sub><sup>-</sup>. *J. Chem. Phys.* 88, 5561 (1988).
20. Liu, Y., Suess, L. & Dunning, F. B. Rydberg electron transfer to SF<sub>6</sub>: Product ion lifetimes. *J. Chem. Phys.* 122, 214313 (2005).
21. Rajput, J., Lammich, L. & Andersen, L. H. Measured Lifetime of SF<sub>6</sub><sup>-</sup>. *Phys. Rev. Lett.* 100, 153001 (2008).
22. Borrelli, R. Franck – Condon analysis of the SF<sub>6</sub><sup>-</sup> electron photodetachment spectrum. *Chem. Phys. Lett.* 445, 84–88 (2007).
23. Gutsev, G. & Bartlett, R. Adiabatic electron affinities of PF<sub>5</sub> and SF<sub>6</sub>: a coupled-cluster study. (1997).
24. Bopp, J. C. *et al.* Spectroscopic Characterization of the Isolated SF<sub>6</sub><sup>-</sup> and C<sub>4</sub>F<sub>8</sub><sup>-</sup> Anions: Observation of Very Long Harmonic Progressions in Symmetric Deformation Modes upon Photodetachment. *J. Phys. Chem. A* 111, 1214–1221 (2007).
25. McMillen, D. F. & Golden, D. M. Hydrocarbon Bond Dissociation Energies. *Annu. Rev. Phys. Chem.* 33, 493–532 (1982).
26. Wang, Y., Champion, R. L., Doverspike, L. D., Olthoff, J. K. & Van Brunt, R. J. Collisional electron detachment and decomposition cross sections for SF<sub>6</sub><sup>-</sup>, SF<sub>5</sub><sup>-</sup>, and F<sup>-</sup> on SF<sub>6</sub> and rare gas targets. *J. Chem. Phys.* 91, (1989).
27. Fehsenfeld, F. C. Electron Attachment to SF<sub>6</sub>. *J. Chem. Phys.* 53, (1970).
28. Avrillier, S. & Schermann, J.-P. Vibrational excitation of SF<sub>6</sub> by electron impact. *Opt. Commun.* 19, 87–91 (1976).
29. Gutsev, G. L. THE STRUCTURE OF THE SF<sub>6</sub> MOLECULE AND THE SF<sub>6</sub><sup>-</sup> ANION EXCITED STATES G. L. Gutsev. *Inst. Chem. Physics, Russ. Acad. Sci.* 5622, 504–510 (1992).

30. Mock, R. S. & Grimsrud, E. P. Electron photodetachment of the molecular anions of SF<sub>6</sub> and several perfluorinated hydrocarbons. *Chem. Phys. Lett.* 184, 99–101 (1991).
31. Ingólfsson, O. & Wodtke, A. Bridging the Cluster-to-Bulk Divide: Electron Attachment Time-of-Flight Mass Spectrometry Reveals Geometrical Shell Closings in (SF<sub>6</sub>)<sub>n</sub> Clusters (n=2 – 550). *Phys. Rev. Lett.* 87, 183401 (2001).
32. Ingólfsson, O. & Wodtke, A. M. Electron attachment time-of-flight mass spectrometry reveals geometrical shell closings in van der Waals aggregates. *J. Chem. Phys.* 117, 3721 (2002).
33. Haberland, H. *et al.* Electronic and Geometric Structure of Ar<sub>n</sub><sup>+</sup> and Xe<sub>n</sub><sup>+</sup> Clusters: The Solvation of Rare-Gas Ions by Their Parent Atoms. *Phys. Rev. Lett.* 67, 3290–3293 (1991).
34. Gascón, J. a., Hall, R. W., Ludewigt, C. & Haberland, H. Structure of Xe<sub>n</sub><sup>+</sup> clusters (N=3–30): Simulation and experiment. *J. Chem. Phys.* 117, 8391 (2002).
35. Even, U., Jortner, J., Noy, D., Lavie, N. & Cossart-Magos, C. Cooling of large molecules below 1K and He clusters formation. *J. Chem. Phys.* 112, 8068 (2000).
36. Zajfman, D. *et al.* Electrostatic bottle for long-time storage of fast ion beams. *Phys. Rev. A* 55, R1577–R1580 (1997).
37. Strasser, D. *et al.* Negative Mass Instability for Interacting Particles in a 1D Box: Theory and Application. *Phys. Rev. Lett.* 89, 283204 (2002).
38. Pedersen, H. *et al.* Diffusion and synchronization in an ion-trap resonator. *Phys. Rev. A* 65, 042704 (2002).
39. Andersen, L. H., Heber, O. & Zajfman, D. Physics with electrostatic rings and traps. *J. Phys. B At. Mol. Opt. Phys.* 37, R57–R88 (2004).
40. Toker, Y. *et al.* The kick-out mass selection technique for ions stored in an Electrostatic Ion Beam Trap. *J. Instrum.* 4, P09001–P09001 (2009).
41. Wu, S., Blake, G. a, Sun, Z. & Ling, J. Simple, high-performance type II b-BaB2O4 optical parametric oscillator. *Appl. Opt.* 36, 5898–901 (1997).
42. Rahinov, I., Toker, Y., Heber, O., Strasser, D. & Rappaport, M. Lifetime measurements in an electrostatic ion beam trap using image charge monitoring. *Rev. Sci. Instrum.* 83, 033302 (2012).
43. WIZA, J. L. Microchannel plate detectors. *Nucl. INSTRUMENTS METHODS* 162, 587–601 (1979).
44. Matsuura, S., Umebayashi, S., Okuyama, C. & Oba, K. Characteristics of the Newly Developed MCP and Its Assembly. *IEEE Trans. Nucl. Sci.* 32, 350–354 (1985).

45. Heber, O. *et al.* Electrostatic ion beam trap for electron collision studies. *Rev. Sci. Instrum.* 76, 013104 (2005).
46. Naaman, R. & Vager, Z. An electron multiplier capable of working at low vacuum: The microsphere plate. *Rev. Sci. Instrum.* 67, 3332 (1996).
47. Toker, Y. *et al.* Blackbody-induced radiative dissociation of cationic SF<sub>6</sub> clusters. *Phys. Rev. A* 86, 023202 (2012).
48. Attia, D., Strasser, D., Heber, O., Rappaport, M. L. & Zajfman, D. Transverse kinematics of ions stored in an electrostatic ion beam trap. *Nucl. Instruments Methods Phys. Res. Sect. A Accel. Spectrometers, Detect. Assoc. Equip.* 547, 279–286 (2005).
49. Straub, H. C., Mangan, M. a., Lindsay, B. G., Smith, K. a. & Stebbings, R. F. Absolute detection efficiency of a microchannel plate detector for kilo-electron volt energy ions. *Rev. Sci. Instrum.* 70, 4238 (1999).
50. Strasser, D. *et al.* Tunable wavelength soft photoionization of ionic liquid vapors. *J. Phys. Chem. A* 114, 879–83 (2010).

## מבוא

בעבודה זו האניון  $SF_6^-$  וצברים קטנים של אניוני  $(SF_6)_n^-$  נבררים על ידי מסה ונכלאים במלכודת יונים אלקטרוסטטית על מנת להחקר עם לייזר ננו-שניות באורך גל משתנה. עקומות יעילות לתלישת אלקטרון נבנו על ידי מדידת פוטו-תוצרים נייטרלים ומדידת מספר היונים הכלואים כפונקציה של אורך גל. מאז שהופיעו השיטות של מדידת פוטו-תוצרים נייטרלים של קרן יונים מהירה עם קרן לייזר קוליניארית בחפיפה טובה וכליאת יונים אלקטרוסטטית, הרחבנו את הטווח עקומות יעילות לתלישת אלקטרון לטווח של 420-700 ננומטר שנמצא מתחת לגבול המדידה של השיטה הקודמת שנוסתה. הצורה של עקומת יעילות לתלישת אלקטרון של  $SF_6^-$  נראית מונוטונית עולה בהתאמה עם המידע שהתקבל ממדידת העקומה הספקטרוסקופית הפוטו-אלקטרונית, דבר המצביע על כך שלא התבצע מעבר עקיף במהלך תלישת האלקטרון בטווח האורכי גל שנמדד. השוואה ישירה של עקומות היעילות לתלישת אלקטרון עבור האניון  $SF_6^-$  וצברים קטנים של אניוני  $(SF_6)_n^-$  שנוצרים על ידי הוספה של אחד או שני מולקולות של  $SF_6$ , מוצאת שהאלקטרון ממקום על אחד ממולקולות ה- $SF_6$  שבצביר. אנחנו גם מתארים ספקטרומטר של פוטו-תוצרים אניונים, שתוכנן, נבנה והותקן בתוך המלכודת יונים האלקטרוסטטית כחלק מעבודת המוסמך שלי. בעתיד רכיב זה יוכל למדוד את יחס התוצרים האניונים השונים שנוצרים בתהליך בליעת הפוטון בנוסף למדידת עקומות יעילות פוטו-ניטרלית.



הפקולטה למתמטיקה ולמדעי הטבע,  
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המחלקה לפיסיקה יישומית

# חתך פעולה לתלישת אלקטרון של $SF_6^-$ ושל צבירי אניוני $(SF_6)_n^-$ קטנים

עבודת מחקר לשם קבלת תואר מוסמך  
בפיסיקה יישומית

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